

SANS under high pressure overview and recent developments of pressure cells used at the ILL

SCHWEINS Ralf

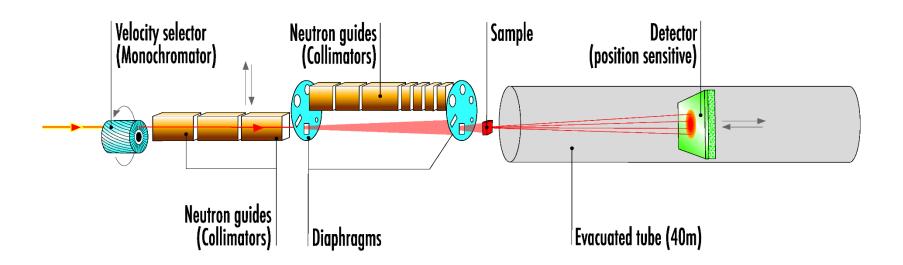
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CS 20 156
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Small Angle Neutron Scattering



The ILL operates 3 SANS instruments: D11, D22, D33

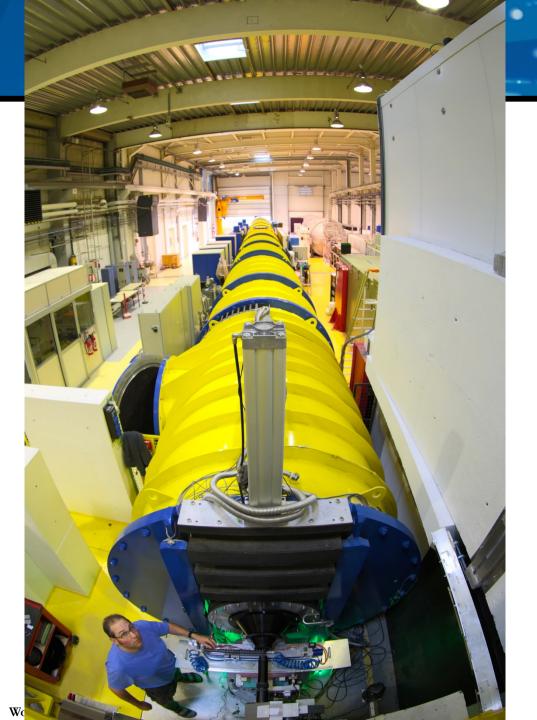
Layout of D11



$$2 \cdot 10^{-4} \text{ Å}^{-1} < q < 0.8 \text{ Å}^{-1}$$







Acknowledgments



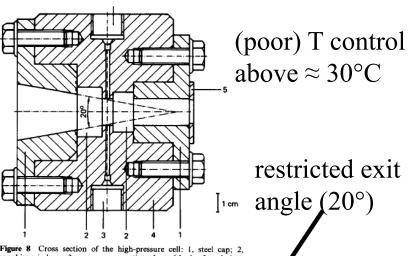
The ILL's high pressure group: C. Payre, J. Maurice, J. Gonthier

- E. Lelièvre-Berna (ILL)
- D. Bowyer, P. Lindner (ILL)
- B. Annighöfer (LLB)
- T. Sottmann and the members of the Strey group (U Cologne)
- J. Peters et al. (hAChE project) (U Grenoble & ILL)
- A. Gelissen, A. Steinschulte et al. (W. Richtering, F. Plamper and groups, RWTH Aachen)

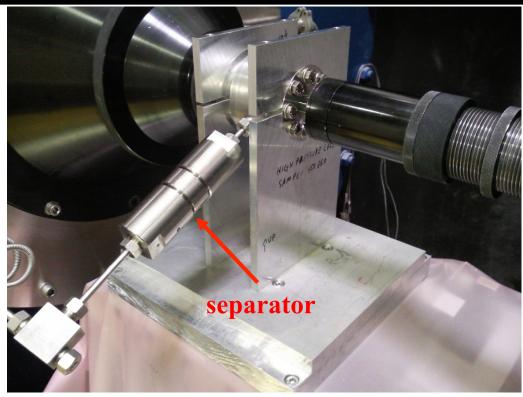
The NMI3 WP20 program for (partial) funding and the discussion meetings

M. Lechner, K. Ibel (U Osnabrück, ILL) Since mid 80's









Purchased from Lechner's workshop for liquid samples, hydrostatic pressure up to p = 2.5 kbar electrical heating jacket -> $T \approx 250$ °C sapphire windows $(2 \times 12 \text{mm}) + 4$ gold sealings sample thickness 4.5mm, optional 2.5mm

M. Lechner, K. Ibel (U Osnabrück, ILL)

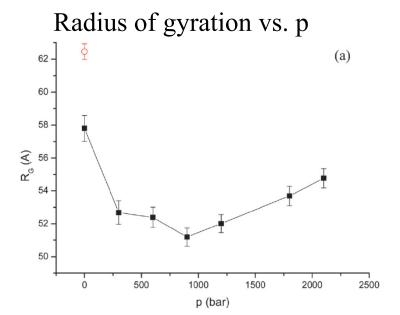


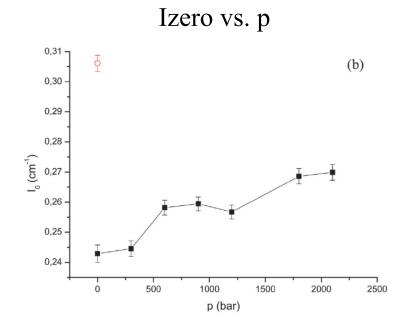


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Pressure-induced molten globule state of human acetylcholinesterase: structural and dynamical changes monitored by neutron scattering†

J. Marion,‡^{ab} M. Trovaslet,‡^{ac} N. Martinez,^{ab} P. Masson,^{ad} R. Schweins,^b F. Nachon,^{ab} M. Trapp^{ef} and J. Peters*^{ab}





hAChE is a mixture of monomers, dimers, tetramers compression until 900 bar, then expansion due to (partial) unfolding

M. Lechner, K. Ibel (U Osnabrück, ILL)





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Molten globule

intermediate state characterised by ...
absence of tertiary structure,
but secondary structure still present to a
great extent

Rg increase between 10 and 30 %

EINS data from IN13

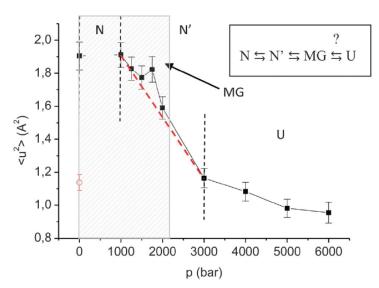


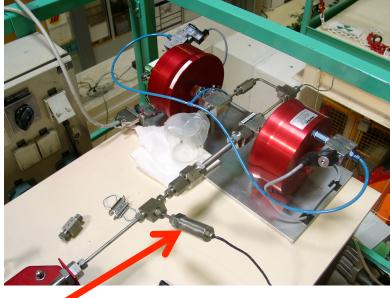
Fig. 4 Mean square displacement of hAChE, measured on IN13, at different pressures up to 6 kbar. The red point (open circle) corresponds to the MSD after releasing pressure. The red dashed line is guidance to the eye to illustrate the straight slope of this part of the curve. The dashed region corresponds to the pressure domain covered by the SANS measurement. The sketch on the right side illustrates the 4-step-model, N being the native, N' the transitional intermediates, MG the molten globule and U the unfolded state.

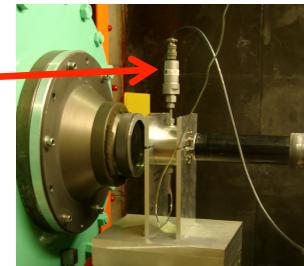
Pressure jumps @ D11 (2003/04)





pressure sensors start of data acquisition via TTL of pressure sensor





used with Lechner / Ibel cell (p < 2500 bar)

hand pump to build up pressure

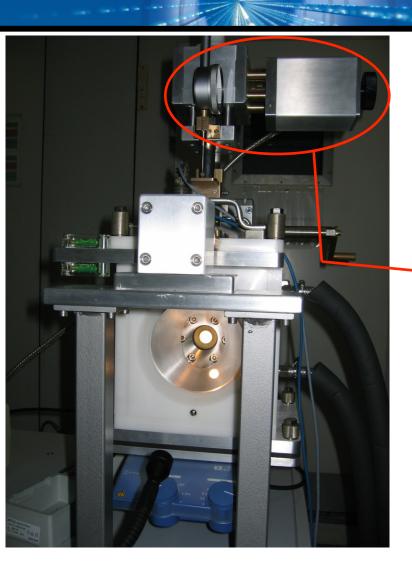
PC controlled automatic valves

PC readable pressure sensors

(initial) dead time 100 msec

Cologne pressure cell





Strey, Sottmann (U Cologne) & D11

Design discussion in 2006, construction done in the university workshop at the university Cologne first experiment on D11 in spring 2007 Sample can be homogenized with magnetic stirrer

Characteristics:

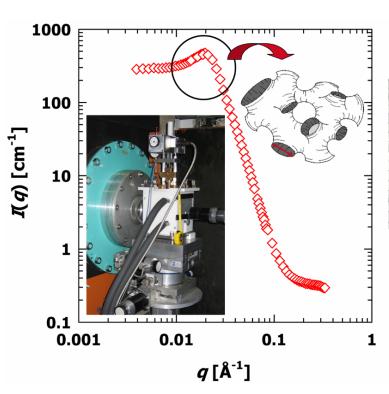
Pressure applied via a plunger / stepper motor !!! no pressure pump & separator needed !!! p < 300 bar

2 sapphire windows (2 x 12mm) Sample volume \approx 18 mL sample thickness 2mm highly precise (\pm 0.05 °C) T-control 10°C < T < 80 °C

weight: 30 kg (!)

Cologne pressure cell





Filling done from the bottom by turning the cell upside down

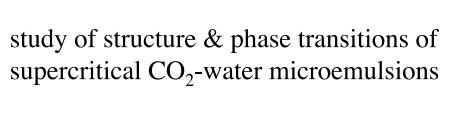


sapphire windows (2 x 12mm)



stainless steel inner core





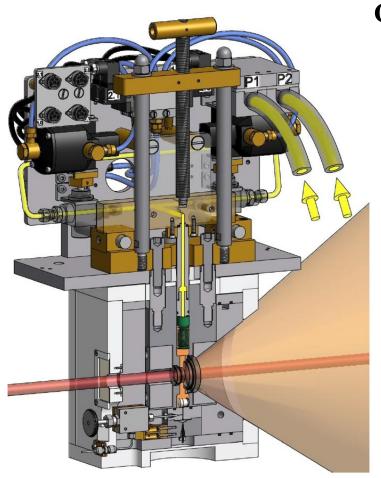


Photos taken from PhD thesis of L. Kramer (U Cologne)

Cologne stroboscopic pressure cell

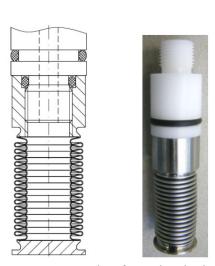


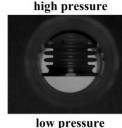
Developed for the TISANE option on D22

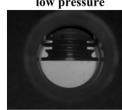


Characteristics:

metal bellow reaching into the sample volume valve circuit to switch between pressures sample volume $\approx 18 \text{ mL}$ $10 \,^{\circ}\text{C} < \text{T} < 50 \,^{\circ}\text{C} \ 0.1 \,^{\circ}\text{C} \ \text{T}$ precision $p_{\text{max}} \ 350 \ \text{bar}$ highest possible p cycle frequency $10 \ \text{Hz}$







taken from PhD thesis of Y. Pütz (U Cologne)

Cologne stroboscopic pressure cell



Developed for the TISANE option on D22

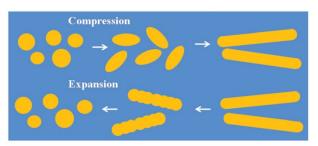


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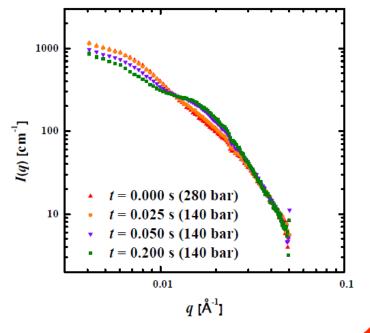
Kinetics of pressure induced structural changes in super- or near-critical CO₂-microemulsions†

Alexander Müller,^a Yvonne Pütz,^a Roland Oberhoffer,^a Nils Becker,^a Reinhard Strey,^a Albrecht Wiedenmann^b and Thomas Sottmann*^c



Scheme 2 Schematical illustration of the two different mechanisms of structural reorganization for compression and expansion.

CO₂-microemulsions show strong pressure dependent properties. Using time-resolved SANS to investigate the kinetics of structural changes upon periodic pressure jumps of adjustable amplitude, we found that the compression-induced formation of cylinders occurs on a timescale of one second, whereas the expansion-induced disintegration into CO₂ swollen spherical micelles is much faster.





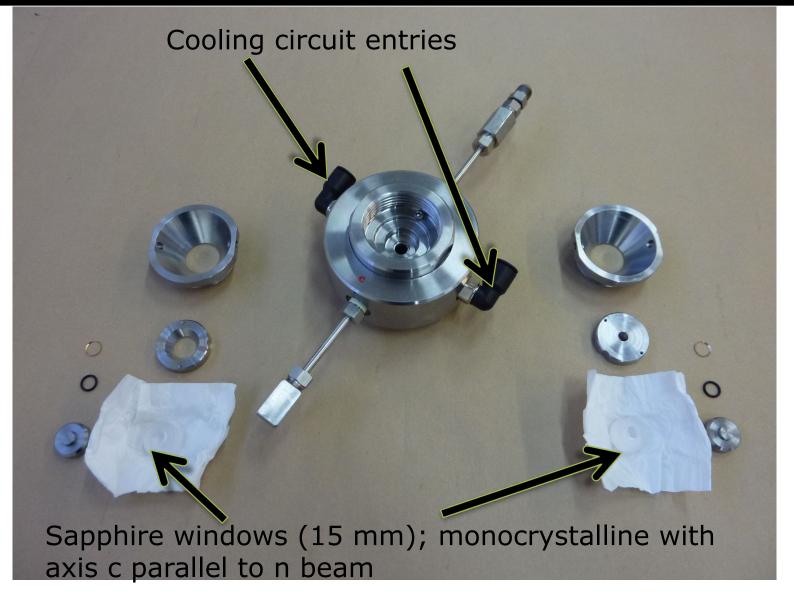




taken from PhD thesis of Y. Pütz (U Cologne)

Nova Swiss 5 kbar hp cell





Nova Swiss 5 kbar hp cell



(Nominal) specifications:

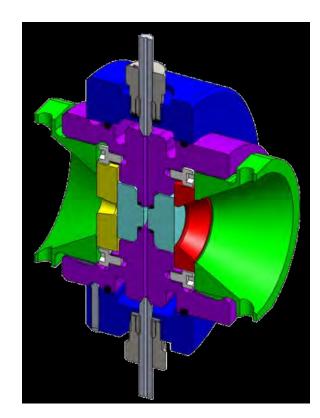
 p_{max} : 5 kbar continuous

T-control with ≈ 1.5 °C precision

T-range: 10 °C < T < 100 °C

sample thickness: 2mm

variable exit angle of 35°, 45°, 63°



Nova Swiss 5 (?) kbar hp cell



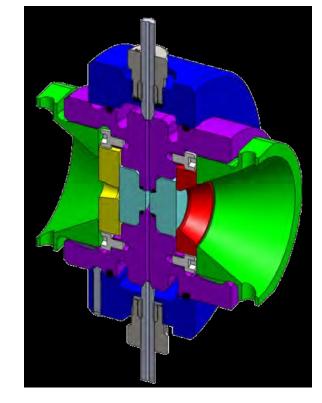
The bitter truth:

- Ø 12 mm incident, exit cone 35°: incident window broken at 3.2 kbar
- Ø 9.5 mm incident, exit cone 35°: incident window broken at 2.3 kbar
- Ø 6 mm incident, exit cone 35°: exit window broken at 2.5 kbar

variable exit aligie of 33,43,03

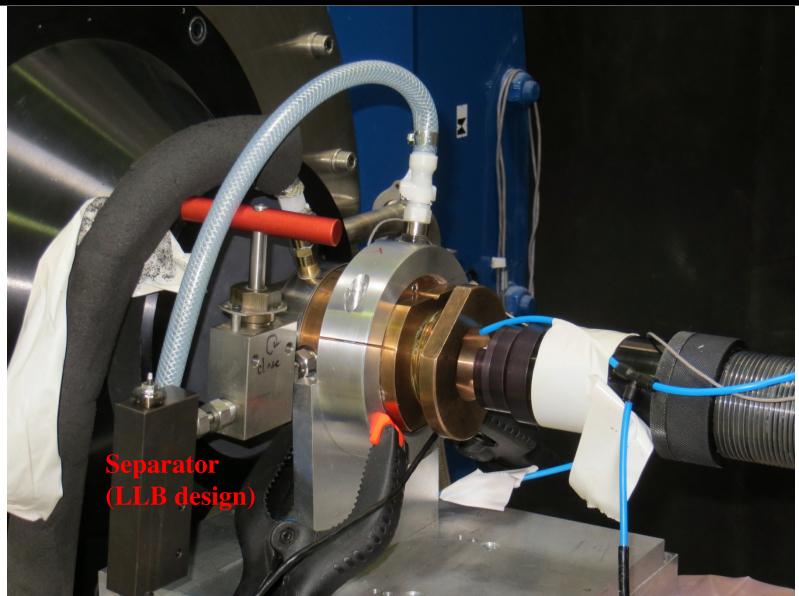






15







Cell body made out of CuBe (common cell for NSE & SANS):

p_{max}: 5 kbar

T-control with ≈ 1.5 °C precision

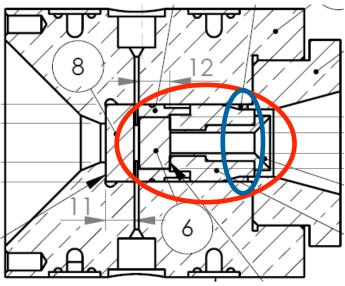
T-range: 5 °C < T < 100 °C

(initial) sample thickness: 2mm

exit angle for 2 kbar: 50°

5 kbar: 35°







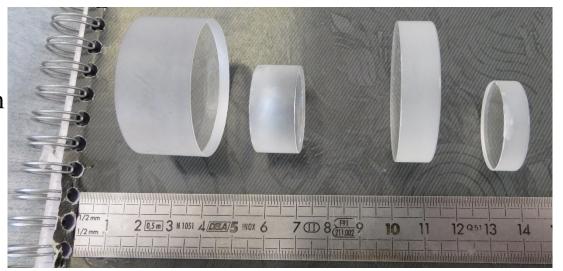
5 kbar

Sets of sapphire windows

2 kbar

5 kbar entrance d 13 mm exit d 20 mm

Tr@ 6Å 84 %



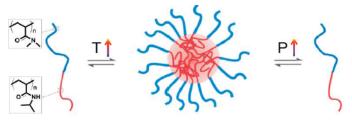
2 kbar entrance d 7 mm exit d 10 mm Tr@ 6Å 89 %



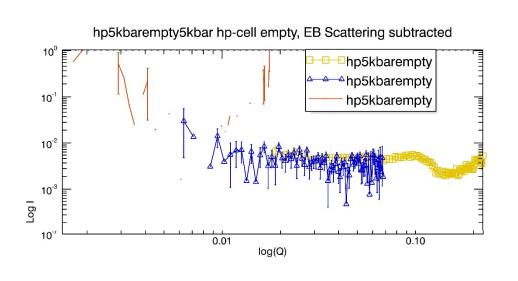


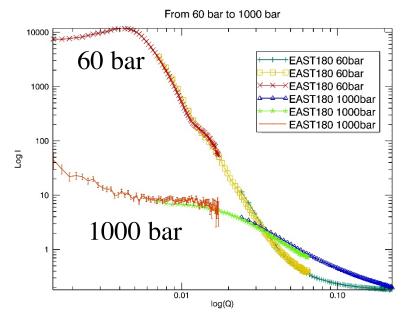
Pressure-induced reverse micellisation of PNIPAM based micelles

groups of Plamper and Richtering (RWTH Aachen)



Scheme 1: Diblock copolymers are present unimolecularly at ambient pressure and below LCST; at temperatures above the LCST micellization occurs and it is supposed that the micelle dissolves again upon pressure application (pressure acts antagonistically to temperature)









Thank you for your attention!